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balanced: 2Ag(s) + Cu2 + (aq) \u2192 2Ag + (aq) + Cu(s) E \u2296 = E \u2296 ox(Ag) + E \u2296 red(Cu2 +) = - 0.80V + 0.34V = - 0.46V. Not spontaneous. ox: [Ni(s) \u2192 Ni2 + + 2e -] \u00d7 5; red: [5e - + 8H + + MnO - 4 \u2192 Mn2 + + 4H 2O] \u00d7 2. balanced: 5Ni(s) + 16H + (aq) + 2MnO - 4 (aq) \u2192 5Ni2 + (aq) + 2Mn2 + (aq) + 8H 2O.

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~~Chapter 18 Nuclear Chemistry~~

a) the energy given off when reactants collide b) *an energy barrier between reactants and products the heat released in a reaction* c) generally very high for a reaction that takes place rapidly

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Chapter 18 Amines and Amides. 18.1 Multiple Choice Questions. 1) The compound CH3 - CH2 - NH - CH3 is classified as a. A) primary amine. B) secondary amine. C) tertiary amine. D) quaternary amine. E) hydrated amine. Answer: B. Objective: 18.1. Global Outcomes: GO2. 2) The compound CH3 - CH2 - NH2 is classified as a. A) primary amine. B) secondary amine.

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