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Remember, the products of this reaction are CO_2 and H_2O .

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Relationship between Mole \u0026
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Gas (# of moles of gas is constant)

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Answer # of moles of Gas in container
(mol) Volume of Container (L)
Temperature of Gas (K) Pressure of
Gas (kPa) A 0.3008 14.0 672.0 120.0
B 0.3008 16.0 1280 200.0 C 0.3008
10.0 340.0 85.0 Table 2: Experimental
Data for Three Different Samples of
Gases That Contain Different Amounts

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of Gas (# of moles of gas is ...

Information: The Gas Constant
And Moles chemquest-36-gases-and-
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3. On planet X, 2.78 moles of a gas

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Answers
takes up 1.85 L under a pressure of 74.1 kPa and a temperature of 201°C. What is the value of the ideal gas constant (R) on planet X? (include units) 0.104 kPa-L/(mol-K) 4. At a pressure of 103 kPa and a temperature of 22°C, 52.9 g of a certain gas has a volume of 31.5 L.

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Answers
What is the identity of this gas?

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mass of the gas and match it with the
periodic table.) Molar mass = 39.97
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a volume of 41.0 L under a pressure of
245 kPa and a temperature of 279 K.

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