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Substituting and multiplying by the factor 109 for the density unity kg/km^3 , the mass of the atmosphere is determined to be $m = 5.092 \times 10^{18}$ kg Discussion Performing the analysis with excel would yield exactly the same results.

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EES Solution for final result: $a=1.2025166$ $b=-0.10167$ $c=0.0022375$ $r=6377$ $h=25$ $m=4 \cdot \pi \cdot (a \cdot r^2 \cdot h + r \cdot (2 \cdot a + b \cdot r) \cdot h^2 / 2 + (a + 2 \cdot b \cdot r + c \cdot r^2) \cdot h^3 / 3 + (b + 2 \cdot c \cdot r) \cdot h^4 / 4 + c \cdot h^5 / 5) \cdot 1E+9$ 1-7 Pressure, Manometer, and Barometer 1-34C The pressure relative to the atmospheric pressure is called ...

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Solution The volume and the weight of a fluid are given. Its mass and density are to be determined. Analysis Knowing the weight, the mass and the density of the fluid are determined to be 3 2 225 N 1 kg m/s 9.80 m/s 1 N W m g ????? ? == =?? ?? ??? ? 23.0 kg 23.0 kg 24 L m ?== =0.957 kg/L V

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Chapter 2 Properties of Fluids 2-7 Solution.
The pressure in a container that is filled with air is to be determined. Assumptions. At specified conditions, air behaves as an ideal

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gas.

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The results are: $\rho(z) = a + bz + cz^2 = 1.20252 - 0.101674z + 0.0022375z^2$ for the unit of kg/m^3 , (or, $\rho(z) = (1.20252 - 0.101674z + 0.0022375z^2) \times 10^9$ for the unit of kg/km^3) where z is the vertical distance from the earth surface at sea level. At $z = 7 \text{ km}$, the equation gives $\rho = 0.600 \text{ kg/m}^3$.

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