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4.1 Intermolecular and interatomic forces (ESBMM) Intermolecular forces.

Intermolecular forces are forces that act between molecules. You will also recall from the previous chapter, that we can describe molecules as being either polar or non-polar. A polar molecule is one in which there is a difference in

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electronegativity between the atoms in the molecule, such that the shared electron pair ...

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bonding, ionic bonding and metallic bonding. In Grade 11, learners will have already studied atomic combinations and molecular structure in more detail.

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demonstrations of all the properties of substances related to their intermolecular forces that were discussed in the last lesson.

06 Practical Demonstrations of
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An experiment for informal assessment is

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included in this chapter. This experiment is very similar to the one on intermolecular forces. In this experiment learners focus on the properties of water. This is a good experiment to do to guide learners in understanding the properties of water.

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The chemistry of water | Intermolecular forces | Siyavula

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The following five experiments investigate the effect of various physical properties (evaporation, surface tension, solubility, boiling point and capillarity) of substances and determine how these properties relate to intermolecular forces. Each experiment will look at a different property. Part 1:

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