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The total m_s is then $5 \times (+1/2) + (?1/2) = 2$. (c) The five electrons with $m_s = +1/2$ use up all of the possible m_l values (+2, +1, 0, ?1, ?2), which sum to zero. The sixth electron has a maximum m_l of +2, so the maximum total m_l is +2. (d) The next available level for one of the 3 electrons is 4p. The remaining five 3d.

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