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The name of a base is the name of the cation followed by the name of the anion (hydroxide). The formula of a base is written by showing the number of hydroxide ions needed to balance the positive charge on the cation. After reading Lesson 9.4, answer the following questions. Names and Formulas of Acids hydrogen1.

### Chemical Names and Formulas - Mr. Mutic's Chemistry ...

a.potassium sulfidee.hydrobromic acidi.sulfur hexafluoride. b.tin(IV) chloridef.aluminum fluoridej.magnesium chloride. c.hydrosulfuric acidg.dinitrogen pentoxidek.phosphoric acid. d.calcium oxideh.iron(III) carbonateI.nitric acid. Name

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charges are anions. The names of nonmetallic anions end in -ide. Metallic cations take the name of the metal. Some metals, including transition metals, can form more than one cation. Use a Roman number in the Stock name or use the classical name with a suffix to name these metals. 1

## Chemical Names and Formulas - Pittsfield

This is a list of the 118 chemical elements which have been identified as of 2020. A chemical element, often simply called an element, is a species of atoms which all have the same number of protons in their atomic nuclei (i.e., the same atomic number, or  $Z$ ).. A popular visualization of all 118 elements is the periodic table of the elements, a convenient tabular arrangement of the elements by ...

## List of chemical elements - Wikipedia

PEARSON CHEMISTRY: The law of conservation of mass is explained in Lesson 2.4 (p. 50). Students are introduced to balancing chemical equations in Lesson 11.1 (pp.

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346- 354). Students learn to write and balance the five different types of chemical reactions in Lesson 11.2 (pp. 356-367). Students learn

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