

Physics Paper 2 HL Markscheme

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Subject Details: Physics HL Paper 2 Markscheme. Candidates are required to answer all questions. Maximum total = 90 marks. 1. Each row in the “Question” column relates to the smallest subpart of the question. 2. The maximum mark for each question subpart is indicated in the “Total” column. 3.

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HL Paper 2 Markscheme In beta minus (??) decay a d quark decays into a u quark, an electron and an electron antineutrino. a. Show that lepton number is conserved in this decay. [1] A nucleus of phosphorus-32 decays by beta minus (?) decay into a nucleus of sulfur-32 . The binding energy per nucleon of is 8.398 MeV and for it is 8.450 MeV.

HL Paper 2 - The Online Physics Tutor

HL Paper 2 Markscheme This question is in two parts. Part 1 is about gravitational force fields. Part 2 is about properties of a gas. Part 1 Gravitational force fields a. State Newton’s universal law of gravitation. [2]

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HL Paper 2 Markscheme Rhodium-106 () decays into palladium-106 () by beta minus (?) decay. The diagram shows some of the nuclear energy levels of rhodium-106 and palladium-106. The arrow represents the ? decay. – – b. Bohr modi?ed the Rutherford model by introducing the condition $mvr = n \cdot h$. Outline the reason for this modi?cation. [3]

HL Paper 2 - The Online Physics Tutor

HL Paper 2. HL Paper 2 Markscheme Examiners report. This question is about an ideal gas. a. Describe how the ideal gas constant R is de?ned. [2] b. Calculate the temperature of 0.100 mol of an ideal gas kept in a cylinder of volume $1.40 \times 10^{-3} \text{m}^3$ at a pressure of $2.32 \times 10^5 \text{Pa}$. [1] The gas in (b) is kept in the cylinder by a freely moving piston. The gas is now heated at constant pressure until the volume occupied by the gas is $3.60 \times 10^{-3} \text{m}^3$.–3 The increase in internal energy of the gas is 760 J.

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