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The Water and Aqueous Systems chapter of this Prentice Hall Chemistry Companion Course helps students learn the essential lessons associated with water and aqueous systems.

Prentice Hall Chemistry Chapter 15: Water and Aqueous ...

Key Concepts 15.1 The high surface tension of water and low vapour pressure are due to the hydrogen bonding between the molecules The structure of ice is a regular open framework of water molecules held together by hydrogen bonds and arranged like a honeycomb Molecules in liquid water are packed closer together than they are in ice

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KOrriban. Prentice Hall Chemistry: Chapter 15. Surface Tension. Surfactant. Key Concepts 15.1. Aqueous Solution. The inward force that tends to minimize the surface area of a.... Any substance that interferes with the hydrogen bonding betwee.... The high surface tension of water and low vapour pressure are....

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Chapter 15 Chemical Equilibrium

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