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Solution Basis: The given reaction By inspection, the carbon balance gives $b = 6$, the hydrogen balance gives $c = 6$, and an oxygen balance $6 + 2a = 6 * 2 + 6$ Gives $a = 6$. Therefore, the balanced equation is Example 16 In the combustion of heptane, CO_2 is produced.

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the solution is the same here as in Problem 2.3, but here the value for the Lagrangian multiplier is the reciprocal of the value in Problem 2.3. 2.5 a. The height of the ball is given by $2 f t() = ?0.5gt + 40 .t$ The value of t for which height is maximized is found by using the first-order condition: $df dt = gt? + = 40 0$, implying $* t = 40 g$. b.

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