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MANUAL 3. Apply How many

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atoms of hydrogen-1 are in a 1.007-g sample? Recall that 1.007 amu is the mass of one atom of hydrogen-1. Round your answer to two significant digits. 1.007 g
1 H-1 atom \times 1.672 $\times 10^{-24}$ g
a. 6.023 $\times 10^{23}$ H-1 atoms 4.

The Mole

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SOLUTIONS MANUAL 30. Compare and contrast sublimation and evaporation. In both processes, the substances become a vapor. During sublimation, the substance goes from the solid phase directly to the vapor phase. During evaporation,

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particles in a liquid gain
enough

States of Matter

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• Chapter 9 141 Section 9.1

Reactions and Equations

pages 282–288 Practice

Problems pages 284–287 Write

skeleton equations for the

following word equations. 1.

Hydrogen and bromine gases

react to yield hydrogen

bromide.

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